

QP CODE: 20100031



Reg No	:	
Name	:	

# **BSc DEGREE (CBCS) EXAMINATION, FEBRUARY 2020**

### Fifth Semester

# Core Course - CH5CRT06 - ORGANIC CHEMISTRY-III

B.Sc Chemistry Model I ,B.Sc Chemistry Model II Industrial Chemistry ,B.Sc Chemistry Model III

Petrochemicals

2017 Admission Onwards

## A99ED2C3

Time: 3 Hours Maximum Marks :60

#### Part A

Answer any **ten** questions.

Each question carries 1 mark.

- 1. What is Borsche's reagent?
- 2. Give an example for phase transfer catalyst.
- 3. Arrange in the order of increasing basic strength: (CH<sub>3</sub>)<sub>3</sub>N, NH<sub>3</sub>, CH<sub>3</sub>NH<sub>2</sub> and (CH<sub>3</sub>)<sub>2</sub>NH.
- 4.  $N_2$ CHCOOEt + HCl  $\rightarrow$  ?
- 5. Represent the following compounds in the increasing order of their aromatic character. Thiophene,Furan,Pyrrole
- 6. Give the structure of an enamine.
- 7. Write the chemical equation for the reaction between glucose and Fehling solution.
- 8. Name the polysaccharide which yields  $\beta$ -glucose upon hydrolysis.
- 9. Give the name and structure of any two suplha drugs.
- 10. What are stimulants? Give one example
- 11 What is an Azo Dye?
- 12. Distinguish between Homopolymers and Heteropolymers.

 $(10 \times 1 = 10)$ 

## Part B

Answer any **six** questions.

Each question carries 5 marks.

Discuss reductive amination of aldehydes and ketones with mechanism.



Page 1/2 Turn Over



- How will you convert aniline to biphenyl? Give mechanism of final step.
- Explain the electrophilic and nucleophilic substitution reactions of quinoline.
- 16. Explain a method for the preparation of ethylacetoacetate.
- 17. What are osazones? How are they prepared?
- 18. How are the following conversions effected (a) Fructose to Glucose. (b) Glucose to Mannose
- 19 Write briefly on antimalarials and antacids.
- 20. How will you prepare Rosaniline? To which class does this dye belong? What is its use?
- 21. Explain briefly on Environmental hazards and biodegradability of polymers.

 $(6 \times 5 = 30)$ 

#### Part C

Answer any two questions.

Each question carries 10 marks.

- 22. Discuss the mechanism of electrophilic substitution reactions-chlorination, nitration and sulphonation in aniline.
- 23. Discuss the electrophilic and nucleophilic substitution reactions of pyridine. Also give the molecular orbital concept regarding its structure.
- 24. (a) What are disaccharides? Draw the cyclic structure of (i) maltose (ii) cellobiose and mention the monosaccharide units present in it.
  - (b) Biiefly explain the reactions and uses of sucrose.
- 25 Discuss the preparation, structure and applications of the following polymers.
  - (a) PET
  - (b) Phenol -formaldehyde Resin
  - (c) Urea-Formaldehyde Resin

 $(2 \times 10 = 20)$ 

